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# ENANTIOSELECTIVE SYNTHESIS OF A SEX ATTRACTANT PHEROMONE OF THE PINE SCALE MATSUCOCCUS FEYTAUDI 

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#### Abstract

The enantioselective synthesis of (-)-( $2 E, 4 E, 6 R, 10 S$ )-4,6, 10-trimethyl-2,4-dodecadien-7-one [2], a primary sex attractant pheromone of the pine scale Matsucoccus feytaudi, is described.


The critical need for effective strategies to control forestland predation by scale insects has fostered efforts to identify the active pheromones of Matsucoccus pine scales. Matsuone [1] has been identified as the primary sex attractant pheromone of the red pine scale Matsucoccus resinosae (United States) and two related species indigenous to Asia, Matsucoccus matsumurae and Matsucoccus thunbergianae (1). The gross structure and relative stereochemistry of $\mathbf{1}$ have recently been confirmed by total synthesis (2). In 1990, a French group reported the isolation and characterization of the structurally related dienes 2 and 3 from collections of Matsucoccus feytaudi, a scale species endemic to Europe (3). We now report the convergent, enantioselective synthesis of both diastereomeric forms of the $M$. feytaudi pheromone $\mathbf{2}$, confirming the assigned structure and establishing the relative stereochemistry of the natural pheromone as priority reflective (4).

Our approach to diene 2 was based on a convergent scheme in which the coupling of an optically active halide and epoxide ( + )-4 would establish the carbon framework and remore stereochemi-
cal relationship of the individual pheromone diastereomers. Thus, treatment of the previously prepared (2) epoxide ( $2 S, 3 R$ )-4 with the Grignard reagent derived from ( $S$ )-1-bromo-2methylbutane (5) afforded alcohol 5, accompanied by small amounts (ca. 6\%) of an inseparable minor diastereomer, an observation consistent with the optical purity of the starting epoxide [ee $88 \%$ (2)]. All compounds reported herein were fully characterized by ir, ${ }^{1} \mathrm{H} \mathrm{nmr}$ and hrcims analysis. Oxidation of 5 with tetrapropylammonium perruthenate (6) in the presence of $N$-methylmorpholine $N$-oxide afforded a $93 \%$ yield of ketone ( $6 R, 10 S$ )-2, accompanied by the diastereomeric $(6 S, 10 S)$ ketone ( $6 \%$ ). Similarly, addition of ( $R$ )-1-bromo-2-methylbutane to ( $2 S, 3 R$ )-4 afforded alcohol $\mathbf{6}$, which upon oxidation yielded ( $6 R, 10 R$ )2 and the minor ( $6 S, 10 R$ ) ketone product (Scheme 1).

Comparison of the $500 \mathrm{MHz}{ }^{1} \mathrm{H}-\mathrm{nmr}$ spectra of the synthetic diastereomers of 2 to the spectrum of M. feytaudi pheromone kindly provided by Dr. J. Einhorn revealed that synthetic ( $6 R, 10 S$ )2 was identical to the natural pheromone, while the ${ }^{1} \mathrm{H}$-nmr spectrum of $(6 R, 10 R)$ -


1



2


$(2 S, 3 R)-4$



5

( $6 R, 10 S$ )-2 (93\%)
$+(6 S, 10 S)$ diastereomer ( $6 \%$ )


6 $\mathrm{NMO}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, 4 A sieves

( $6 R, 10 R$ )-2 ( $93 \%$ )
$+(6 S, 10 R)$ diastereomer $(6 \%)$

Scheme 1

2 was clearly distinct from that of authentic 2 . Individual diastereomers of $\mathbf{2}$ are easily distinguished by the ${ }^{1} \mathrm{H}-\mathrm{nmr}$ chemical shifts of signals assigned to the diastereotopic $\mathrm{C}-8$ protons. For authentic 2 and ( $6 R, 10 S$ )-2, the $\mathrm{H}-8$ signals appear at 2.19 (ddd, $J=16.8,9.3,5.5 \mathrm{~Hz}$ ) and $2.29 \mathrm{ppm}(\mathrm{ddd}, J=16.8,8.6,6.4 \mathrm{~Hz}$ ), while for ( $6 R, 10 R$ )-2 the $\mathrm{H}-8$ signals are observed at 2.13 (ddd, $J=16.8,8.7$, 6.1 Hz ) and 2.32 ppm (ddd, $J=16.8$, $9.3,5.5 \mathrm{~Hz}$ ). A similar chemical shift pattern is observed for diastereomers of matsuone [1].

In summary, we have completed the first asymmetric synthesis of the major component of the $M$. feytaudi pheromone complex. Our observations serve to confirm the assigned structure of 2 and establish the relative stereochemistry of the natural product as priority reflective. We note that the syn relative configuration of the M. feytaudi pheromone corresponds to that of matsuone [1] (2). Efforts to define the absolute configuration of $\mathbf{2}$ and extend our scheme to the preparation of other Matsucoccus pheromones will be the subject of future reports.

## EXPERIMENTAL

General experimental procedures.Nmr spectra were recorded on General Electric

QE-300 or GN-500 nmr spectrometers. All spectra were recorded using $\mathrm{CDCl}_{3}$ or $\mathrm{C}_{6} \mathrm{D}_{6}$ solutions. ${ }^{1} \mathrm{H}$-nmr spectra were acquired at 300 MHz and 500 MHz , and peak positions are reported in parts per million ( ppm ) with $\mathrm{CHCl}_{3}$ or $\mathrm{C}_{6} \mathrm{H}_{6}$ ( 7.26 ppm or 7.15 ppm respectively) as internal standards. ${ }^{13} \mathrm{C}$-nmr spectra were acquired at 75 MHz , and peak positions are reported in ppm with $\mathrm{CDCl}_{3}$ as internal standard (center peak of triplet at 77 ppm ). Ir spectra were recorded on an IBM instruments IR/32 Ft-ir. Optical rotations were determined on a Perkin-Elmer 241 polarimeter. Low resolution mass spectra were acquired at SUNY College of Environmental Science and Forestry. High resolution mass spectra were performed at the University of Illinois, Ur-bana-Champaign.

Solvents were distilled prior to use except as noted. Preparative chromatography columns and reactions were monitored by tlc utilizing Merck Si gel $60 \mathrm{~F}_{254}$ precoated 0.25 mm plates, developed in hexane- $\mathrm{Et}_{2} \mathrm{O}$ (4:1); $R_{f}$ values are reported for this solvent mixture unless otherwise noted. All reactions were run under an inert atmosphere of argon unless otherwise noted.
(2E,4E,6R,7R,10S)-4,6,10-TRIMETHYL-2,4-DODECADIEN-7-OL [5].-( $2 E, 4 E, 6 R$ )-4-methyl-6-[(1S)-oxiranyl]-2,4-heptadiene [4] (7.0 mg, $46.0 \mu \mathrm{~mol}$ ) was dissolved in THF ( 0.5 ml ) and treated with $\mathrm{CuI}(1.4 \mathrm{mg}, 7.4 \mu \mathrm{~mol})$. To this solution was added ( $2 S$ )-2-methylbutylmagnesium bromide ( $1.00 \mathrm{ml}, 0.66 \mathrm{M}$ in THF, $660 \mu \mathrm{~mol}$ ). After $4 h$ the reaction was quenched with $\mathrm{NH}_{4} \mathrm{Cl}$ ( 1 ml , aqueous saturated), extracted with $\mathrm{Et}_{2} \mathrm{O}$ ( $3 \times 2 \mathrm{ml}$ ), dried, and concentrated ( $75^{\circ}$ at atmospheric pressure) to a volume of 1 ml . The resulting solution was subjected to Cc Ipentane,
pentane- $\mathrm{Et}_{2} \mathrm{O}$ (16:1)] to afford ( $2 E, 4 E, 6 R, 7 R, 10 S$ )-4,6,10-trimethyl-2,4-dodecadien-7-ol [5] (6.2 $\mathrm{mg}, 60 \%$ ) containing ca. $6 \%$ of the ( $6 S, 10 S$ ) diastereomer, as determined by the ${ }^{1} \mathrm{H}$-nmr spectrum: $R_{f} 0.36$ [pentane- $\left.\mathrm{Et}_{2} \mathrm{O}(4: 1)\right] ;[\alpha]^{24} \mathrm{D} 9.4$ ( $c=0.620, \mathrm{CDCl}_{3}$ ) [optical rotation is not corrected for the presence of the $(6 S, 10 S)$ diastereomer]; ir (neat/ NaCl ) 3353, 2961, 2930, 2874, 963; ${ }^{13} \mathrm{C}$ nmr 136.1, 133.9, 132.3, 122.8, 76.4, $38.8,34.5,32.9,32.1,29.7,19.1,18.1,16.3$, $12.9,11.4 ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, 7.26\right)$, $6.08(\mathrm{~d}, J=15.4,1 \mathrm{H}, \mathrm{H}-3), 5.61(\mathrm{dq}, J=15.4$, $6.7,1 \mathrm{H}, \mathrm{H}-2), 5.20(\mathrm{~d}, J=9.7,1 \mathrm{H}, \mathrm{H}-5), 3.39$ (vbrs, 1H, H-7), 2.62-2.50 (m, 1H, H-6), 1.77 (d, $J=6.7,3 \mathrm{H}, \mathrm{H}-1$ ), 1.75 ( $\mathrm{s}, 3 \mathrm{H}, 4-\mathrm{Me}$ ), 1.55-1.02 (m, 7H, H-8, H-9, H-10, H-11), $1.01(\mathrm{~d}, J=6.7,3 \mathrm{H}, 6-\mathrm{Me}), 0.89-0.83(\mathrm{~m}, 6 \mathrm{H}$, $\mathrm{H}-12,10-\mathrm{Me}$ ); eims $\mathrm{m} / \mathrm{z}$ (rel. int.) $[\mathrm{M}]^{+} 224$ (0.56), 110 (49.60), 109 (22.02), 95 (base), 67 (14.49), 55 ( 18.70 ); hrcims ( $\mathrm{CH}_{4}, \mathrm{M}+\mathrm{H}$ ) calcd for $\mathrm{C}_{15} \mathrm{H}_{29} \mathrm{O}, 224.2218$, found 224.2213 .
( $2 E, 4 E, 6 R, 7 R, 10 R$ )-4, 6, 10-TRIMETHYL- $2,4-$ DODECADIEN-7-OL [6].-( $2 E, 4 E, 6 R$ )-4-methyl-6-[(1S)-oxiranyl\}-2,4-heptadiene [4] (7.2 mg, $47.6 \mu \mathrm{~mol})$ was dissolved in THF ( 0.5 ml ) and treated with $\mathrm{CuI}(1.3 \mathrm{mg}, 6.8 \mu \mathrm{~mol})$. To the resulting solution was added ( $2 R$ )-2-methylbutylmagnesium bromide ( $0.50 \mathrm{ml}, 0.66 \mathrm{M}$ in THF, $330 \mu \mathrm{~mol}$ ). After 3 h the reaction was quenched with $\mathrm{NH}_{4} \mathrm{Cl}(1 \mathrm{ml}$, aqueous saturated), extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 2 \mathrm{ml})$, dried, and concentrated ( $65^{\circ}$ at atmospheric pressure) to a volume of 1 ml . The resulting solution was subjected to cc [pentane, pentane- $\mathrm{Et}_{2} \mathrm{O}$ (16:1)] to afford ( $2 E, 4 E, 6 R, 7 R, 10 R$ )-4,6, 10-trimethyl- 2,4 -dodeca-diene-7-ol [6] ( $6.2 \mathrm{mg}, 58 \%$ ) containing ca. $6 \%$ of the $(6 S, 10 R)$ diastereomer as determined by the ${ }^{1} \mathrm{H}$-nmr spectrum: $R_{f} 0.34$ [pentane- $\mathrm{Et}_{2} \mathrm{O}$ (4:1)]; $[\alpha]^{24}$ D $8.8\left(c=0.340, \mathrm{CDCl}_{3}\right)$ [optical rotation is not corrected for the presence of the $(6 S, 10 R)$ diastereomer\}; ir (neat $/ \mathrm{NaCl}$ ) 3401, 2959, 2928, 2874, 963 ; ${ }^{13} \mathrm{C} \mathrm{nmr} \mathrm{136.1}, \mathrm{134.2}$, $122.8,38.7,34.6,32.9,32.2,29.3,19.3$, $18.1,16.1,12.9,11.3 ;{ }^{1} \mathrm{H} \mathrm{nmr}$ ( 300 MHz , $\left.\mathrm{CDCl}_{3}, 7.26\right) 6.08(\mathrm{~d}, J=15.5,1 \mathrm{H}, \mathrm{H}-3), 5.61$ $(\mathrm{dq}, J=15.5,6.6,1 \mathrm{H}, \mathrm{H}-2), 5.21(\mathrm{~d}, J=9.7$, 1H, H-5), 3.42-3.35 (m, 1H, H-7), 2.63-2.51 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H}-6$ ), 1.77 (d, $J=6.6,3 \mathrm{H}, \mathrm{H}-1), 1.75$ (s, $3 \mathrm{H}, 4-\mathrm{Me}$ ), 1.56-1.05 (m, 7H, H-8, H-9, $\mathrm{H}-10, \mathrm{H}-11$ ), 1.05 (d, $J=6.7,3 \mathrm{H}, 6-\mathrm{Me}$ ), $0.88-0.83$ (m, 6H, H-12, C-10 methyl); eims $m / z$ (rel. int.) $[\mathrm{M}]^{+} 224(1.02), 110(55.84), 109$ (22.69), 95 (base), 67 (12.82), 55 (17.67); hrcims $\left(\mathrm{CH}_{4}, \mathrm{M}+\mathrm{H}\right)$ calcd for $\mathrm{C}_{15} \mathrm{H}_{29} \mathrm{O}$, 224.2218, found 224.2216.
( $2 E, 4 E, 6 R, 10 S$ )-4,6, 10-TRIMETHYL-2,4-DO-DECADIEN-7-ONE ( $6 R, 10 S-[2]$ )- ( $2 E, 4 E, 6 R$, $7 R, 10 S)-4,6,10$-Trimethyl-2,4-dodecadien-7-ol [5] ( $3.3 \mathrm{mg}, 14.7 \mu \mathrm{~mol}$ ) was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 0.3 ml ), and a $1 / 8^{\prime \prime} 4 \AA$ sieve was added. The re-
sulting solution was treated successively with 4methylmorpholine 4 -oxide ( $3.0 \mathrm{mg}, 2.1 \mu \mathrm{~mol}$ ) and tetrapropylammonium perruthenate ( 1.0 $\mathrm{mg}, 2.95 \mu \mathrm{~mol}$ ) and stirred 30 min . The reaction mixture was subjected to cc [pentane, pentane$\left.\mathrm{Et}_{2} \mathrm{O}(16: 1)\right]$ to afford quantitatively ( $2 E, 4 E, 6 R$, 10S)-4,6, 10-trimethyl-2,4-dodecadien-7-one [2] $(3.3 \mathrm{mg})$ containing $\mathrm{ca} .6 \%$ of the $(6 S, 10 S)$ diastereomer, as determined by the ${ }^{1} \mathrm{H}-\mathrm{nmr}$ spectrum: $R_{f} 0.80$ [pentane- $\left.\mathrm{Et}_{2} \mathrm{O}(4: 1)\right] ;[\alpha]^{24} \mathrm{D}-48.5$ ( $c=0.330, \mathrm{CDCl}_{3}$ ) [optical rotation is not corrected for the presence of the ( $65,10 S$ ) diastereomer); ir (neat/ NaCl ) 2961, 2928, 2874, 2855, 1717, $963 ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 7.15\right), 6.01$ (dq, $J=15.6,1.5,1 \mathrm{H}, \mathrm{H}-3), 5.49(\mathrm{dqd}, J=15.6$, $6.5,0.5,1 \mathrm{H}, \mathrm{H}-2), 5.24(\mathrm{brd}, J=9.9,1 \mathrm{H}, \mathrm{H}-$ $5), 3.28(\mathrm{dq}, J=9.9,6.8,1 \mathrm{H}, \mathrm{H}-6), 2.26$ (ddd, $J=16.8,9.3,6.3,1 \mathrm{H}, \mathrm{H}-8$ ), 2.15 (ddd, $J=$ $\left.16.8,9.2,5.7,1 \mathrm{H}, \mathrm{H}-8^{\prime}\right), 1.71-1.58(\mathrm{~m}, 1 \mathrm{H}$, $\mathrm{H}-9), 1.66(\mathrm{~d}, J=1.5,3 \mathrm{H}, 4-\mathrm{Me}), 1.60(\mathrm{dd}, J=$ $6.5,1.3,3 \mathrm{H}, \mathrm{H}-1$ ), $1.42-1.14$ (m, 3H, H-9', $\mathrm{H}-10, \mathrm{H}-11$ ), 1.13 (d, $J=6.8,3 \mathrm{H}, 6-\mathrm{Me}$ ), $1.06-1.00\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-11^{\prime}\right), 0.80(\mathrm{t}, J=7.3$, $3 \mathrm{H}, \mathrm{H}-12$ ), 0.75 (d, $J=6.8,3 \mathrm{H}, 10-\mathrm{Me}$ ); eims $m / z$ (rel. int.) $[\mathrm{M}]^{+} 222$ (6.99), 113 (11.53), 109 (base), 95 (21.47), 81 (5.88), 67 (14.98); hrcims $\left(\mathrm{CH}_{4}, \mathrm{M}+\mathrm{H}\right)$ calcd for $\mathrm{C}_{15} \mathrm{H}_{27} \mathrm{O}, 223.2062$, found 223.2052.
( $2 E, 4 E, 6 R, 10 R$ )-4,6, 10-Trimethyl-2,4-DO-DECADIEN-7-ONE ( $6 R, 10 R-[2])$ - ( $2 E, 4 E, 6 R$, $7 R, 10 R$ )-4,6,10-Trimethyl-2,4-dodecadien-7-ol [6] ( $3.1 \mathrm{mg}, 13.8 \mu \mathrm{~mol}$ ) was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 0.3 ml ), and a $1 / 8^{\prime \prime} 4 \AA$ sieve was added. The resulting solution was treated successively with 4methylmorpholine 4 -oxide ( $3.0 \mathrm{mg}, 22.1 \mu \mathrm{~mol}$ ) and tetrapropylammonium perruthenate (1.0 $\mathrm{mg}, 2.95 \mu \mathrm{~mol}$ ) and stirred 30 min . The reaction mixture was subjected to cc [pentane, pentane$\mathrm{Et}_{2} \mathrm{O}(16: 1)$ ) to afford quantitatively ( $2 E, 4 E, 6 R$, 10R)-4,6,10-trimethyl-2,4-dodecadien-7-one [2] $(3.1 \mathrm{mg})$ containing ca. $6 \%$ of the $(6 S, 10 R)$ diastereomer as determined by the ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectrum: $R_{f} 0.80\left\{\right.$ pentane- $\left.\mathrm{Et}_{2} \mathrm{O}(4: 1)\right\} ;[\alpha]^{24} \mathrm{D}-44.0(c=$ $0.150, \mathrm{CDCl}_{3}$ ) [optical rotation is not corrected for the presence of the ( $6 S, 10 \mathrm{R}$ ) diastereomer]; ir (neat/ NaCl) 2961, 2928, 2874, 1717, 963; ${ }^{1} \mathrm{H}$ nmr ( $500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 7.15$ ) 6.01 (br d, $J=$ $15.5,1 \mathrm{H}, \mathrm{H}-3), 5.48(\mathrm{dq}, J=15.5,6.6,1 \mathrm{H}$, $\mathrm{H}-2$ ), 5.24 (br d, $J=9.7,1 \mathrm{H}, \mathrm{H}-5$ ), 3.28 (dq, $J=9.7,6.7,1 \mathrm{H}, \mathrm{H}-6$ ), 2.29 (ddd, $J=16.8$, $9.3,5.9,1 \mathrm{H}, \mathrm{H}-8$ ), 2.11 (ddd, $J=16.8,9.4$, $6.0,1 \mathrm{H}, \mathrm{H}-8^{\prime}$ ), $1.67-1.58$ (m, 1H, H-9), 1.65 (br s, $3 \mathrm{H}, 4-\mathrm{Me}$ ), 1.60 (dd, $J=6.6,1.4,3 \mathrm{H}, \mathrm{H}-$ 1), $1.44-1.15$ ( $\mathrm{m}, 3 \mathrm{H}, \mathrm{H}-9^{\prime}, \mathrm{H}-10, \mathrm{H}-11$ ), $1.13(\mathrm{~d}, J=6.7,3 \mathrm{H}, 6-\mathrm{Me}), 1.06-0.99(\mathrm{~m}, 1 \mathrm{H}$, H-11'), $0.80(\mathrm{t}, J=7.4,3 \mathrm{H} \mathrm{H}-12$ ), 0.75 ( $\mathrm{d}, J=$ $6.5,3 \mathrm{H}, 10-\mathrm{Me}$ ); eims $m / z$ (rel. int.) [M] ${ }^{+} 222$ (6.27), 113 (11.21), 109 (base), 95 (22.49), 81 (6.23), 67 (15.68); hrcims ( $\mathrm{CH}_{4}, \mathrm{M}+\mathrm{H}$ ) calcd for $\mathrm{C}_{15} \mathrm{H}_{27} \mathrm{O}, 223.2062$, found 223.2056 .

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